

Journal of Alloys and Compounds 323-324 (2001) 315-320



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Growth and 10 K spectroscopy of Nd³⁺ in NaBi(WO₄)₂ single crystal

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Abstract

The experimental procedures to grow Nd doped NaBi(WO₄)₂ single crystals by the Czochralski method and the difficulties or limitations encountered for doping are described. Nd concentrations in the crystal up to 0.3×10^{20} cm⁻³ have been produced with optical quality. Polarised optical absorption and photoluminescence of Nd³⁺ ion in this host have been studied at 10 K and the relative energies of 90 levels have been determined. An attempt to explain the large observed spectroscopic line-widths has been conducted through crystal field calculations of these levels supposing either a single Nd position with C₂ symmetry or two centres both with S₄ point symmetry. The latter assumption produces energy differences between both optical centres that can justify the enlargement of lines, but it is not consistent with the OA intensity changes observed in polarised spectra. The coexistence of several sites with statistically modified ordering of the short range Na and Bi cation distribution is suggested to account for this disagreement. © 2001 Elsevier Science B.V. All rights reserved.

Keywords: Phosphors; Crystal growth; Crystal field; Optical properties; Luminescence

1. Introduction

Crystals of double tungstates and molybdates with structure related to the CaWO₄ scheelite-type are particularly promising optical materials [1], whose spectroscopic, laser, and χ^3 non-linear optical properties, especially when doped with Nd³⁺, have been studied [2]. Among them, sodium bismuth tungstate with stoichiometry NaBi(WO₄)₂ (NBW) presents several interesting features, mainly concerning the luminescence properties [3], and the polarised IR and Raman spectra characteristics [4,5]. The relationship between these measurements and short-range order as well as non-random distribution of Na⁺ and Bi³⁺ ions in the crystal has been analysed [6].

Continuing a preliminary crystal growth study on this system [7], the current work presents new results on the growth and the spectroscopic properties of Nd³⁺ doped NBW crystals. From 10 K polarised optical absorption and photoluminescence measurements, the relative energy of 90 crystal field levels has been identified and 85 of them used to perform a single electron crystal field analysis of

the Nd^{3+} centre. The symmetry of the point site occupied by Nd^{3+} when it replaces Bi^{3+} in the NBW matrix as well as the large widths of the observed spectral line profiles are discussed and an attempt to explain them is made.

2. Crystal structure

 $NaBi(WO_4)_2$ is described as belonging to the tetragonal scheelite-type crystal structure. A summary of the crystal chemistry of NBW has been previously given [6]. Nevertheless, some contradictory results about the space group S.G. and the symmetry of the point sites occupied by the metallic cations arise from earlier X-ray studies by the same [5] or different [8] authors. While initially [5,8] the symmetry of the structure was proposed to correspond to the centrosymmetric S.G. $I4_1/a$ (No. 88), Z=2, a more recent single-crystal X-ray analysis [6] described it in the non-centrosymmetric S.G. I4 (No. 82), Z=2. In any case, the structure is built-up with WO₄ tetrahedral groups and NaO₈ and BiO₈ distorted square antiprisms. In the first model Na⁺ and Bi³⁺ ions occupy statistically an unique position, with S_4 [8] or C_2 local site symmetry [5], however, in the non-centrosymmetric model these cations share two non-equivalent lattice sites (2c and 2b Wyckoff positions for Bi1 and Bi2, respectively) with local symme-

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try S₄ but different occupancy factors, namely 60%Bi+ 40%Na and 40%Bi+ 60%Na for 2*c* and 2*b* sites, respectively.

3. Crystal growth

An important step in growing $NaBi_{1-x}Nd_x(WO_4)_2$ single crystals with high optical quality is the synthesis of the correct phase before pulling. This has been achieved by a two-step process. First, the intermediate $Na_2W_4O_{13}$ phase was sintered by 100 h/760°C annealing of 99% WO₃ and 99.5% NaCO₃ stoichiometric mixtures. Further, the Nd doped NBW phase was obtained by reacting this intermediate phase with 99.9% Bi₂O₃ and Nd₂O₃ during 100 h at 860°C.

During sintering and crystal pulling of NBW crystals we observed evaporation and condensation on the seed-holder and crucible walls of the evaporated material. We have determined by ionic coupling plasma (ICP) spectrometry that the composition of the evaporated material is $Na_{0.5}K_{0.5}Bi(MoO_4)_2$. The presence of Mo is likely due to the low melting point of MoO₃ (m.p. 795°C), this oxide being a major contamination of WO₃. The source of K is not clear at present. This evaporation modifies the melt composition during pulling leading to thermal melt instability, non-flat liquid-solid interface, non-uniform crystal diameter and a non-uniform Nd incorporation. The sintering procedure described above improves these aspects and further purification of the starting compounds is expected to improve the final results. Nevertheless it is worth noting that possible Mo or K contamination of the NBW studied crystal is well below the analytical sensitivity and therefore does not influence the spectroscopy of Nd. Further details of the crystal growth procedure as well as of characterization of the sample can be found elsewhere [9].

Nd doped NBW crystals with dimensions of several centimetres were grown using Pt crucibles with pulling rates in the 0.8–4 mm/h range and crystal rotation rates between 6 and 10 rpm. The melting temperature was 935° C. Nd concentrations in the crystal up to $[Nd]=0.3 \times 10^{20}$ cm⁻³ were achieved with good optical quality. With the procedure at present developed, above this Nd concentration crystal cracks, inclusions and even polycrystal-line regions were observed.

4. Optical spectroscopy

Optical absorption (OA) was recorded in a Variant 5E spectrophotometer. Continuous wave photoluminescence (PL) was excited with the ultraviolet multiline emission of a 5 W Ar laser, the emission was analysed with a Spex 340E spectrometer and detected with a cooled R928

Hamamatsu photomultiplier, or a 77 K cooled Ge photodiode and a lock-in amplifier. For OA and PL experiments the samples were cooled to 10 K using a He close cycled cryostat connected to a suitable temperature controller.

Fig. 1 gives an overview of the 10 K polarised OA spectra for Nd³⁺. From these measurements the energies of the Stark levels from ${}^{4}I_{13/2}$ up to ${}^{2}I_{11/2}$ have been determined. J-manifolds of 4f³ configuration of Nd³⁺ are split in (J+1)/2 Stark levels. The presence of a single band from the ground level to the ${}^{2}P_{1/2}$ multiplet would suggest the existence of a unique site for Nd³⁺ in the crystal. This would agree with the assumption of the centrosymmetric crystal structure and, therefore, with a C₂ site symmetry for Nd³⁺, but from this symmetry no polarisation rules are expected for the Nd³⁺ intraionic transitions, in contrast with the results shown in Fig. 1.

It must be noted, however, that the spectral widths of the absorption lines of Nd³⁺ in NBW are in most cases larger than in other crystalline hosts. For instance, the ${}^{4}I_{9/2} \rightarrow {}^{2}P_{1/2}$ transition in NBW has a bandwidth of 21 cm⁻¹ while in most crystals it is $\approx 8-10$ cm⁻¹. The widths of some OA bands, which do not appear strongly overlapped, are included in Table 1. The large band-width could be due to the overlapping of the OA of several centres, therefore, the contribution of the two centres assumed for the non-centrosymmetric crystal structure must be examined. For this purpose in what follows we have made a crystal field (cf) analysis of the Nd³⁺ energy levels in NBW.

Energy level positions of the ground ${}^{4}I_{9/2}$ and first excited ${}^{4}I_{11/2}$ multiplets have been established from the PL lowest ${}^{4}F_{3/2}$ level shown in Fig. 2. Lesser intensity emission bands arising from the ${}^{4}F_{3/2}$ high-energy component, 62 cm⁻¹ above, unambiguously reproduce the sequence of these ${}^{4}I_{9/2}$ and ${}^{4}I_{11/2}$ crystal field levels (Fig. 2). From the OA and PL experiments, the energies of the 90 levels summarised in Table 1 have been determined. For the cf calculations we have excluded the five ${}^{2}H_{11/2}$ levels, which are argued to be affected by a non-negligible mixing with some states of the $4f^{2}5d^{1}$ excited configuration [10], and thus usually poorly reproduced by normal cf calculations.

5. Simulation of the cf energy levels and discussion

The energy of Nd³⁺ levels in solids can be properly simulated supposing the central field approximation which considers separate Hamiltonians for free ion (fi) and crystal field (cf) interactions. The effective free ion Hamiltonian $H_{\rm FI}$ used here for Nd³⁺ includes up to 20 fi parameters [11]. The one-electron crystal field potential $H_{\rm CF}$ is expressed as a sum of products of tensor operators $(C_q^k)_i$, with real B_q^k and complex S_q^k parameters as coefficients [12]:



Fig. 1. Polarised 10 K optical absorption of Nd³⁺ in NaBi(WO₄)₂ single crystal [Nd] = 0.3×10^{20} cm⁻³. α spectra (E \perp c, H \perp c); π spectra (E//c, H \perp c).

$$H_{\rm CF} = \sum_{k=2q=0}^{4,6} \sum_{q=0}^{k} \left[B_q^k (C_q^k + (-1)^q C_{-q}^k) + i S_q^k (C_q^k - (-1)^q C_{-q}^k) \right]$$
(1)

where the values of k and q for which the parameters are non-zero depend on the considered site symmetry.

When the S₄ symmetry is considered for Nd³⁺ in NBW, the cf potential involves five real B_q^k and two complex S_q^k Table 1

10 K observed (E_{o}) and calculated (E_{c}) energy levels (cm⁻¹) of Nd³⁺ in NaBi(WO₄)₂ crystals in a single site of C₂ symmetry and for two possible S₄ symmetry sites. ΔE is the experimental OA line-width

h ₁ 0 11 0 <th>$^{2S+1}L_{J} \\$</th> <th>$E_{\rm o}~(\Delta E)$</th> <th>E_{c}-C₂</th> <th>$E_{\rm c}$1-S₄</th> <th>$E_{\rm c}$2-S₄</th> <th>$^{2S+1}L_{J}$</th> <th>$E_{\rm o}~(\Delta E)$</th> <th>E_{c}-C₂</th> <th>$E_{c}1-S_{4}$</th> <th>$E_{\rm c}$2-S₄</th>	$^{2S+1}L_{J} \\$	$E_{\rm o}~(\Delta E)$	E_{c} -C ₂	$E_{\rm c}$ 1-S ₄	$E_{\rm c}$ 2-S ₄	$^{2S+1}L_{J}$	$E_{\rm o}~(\Delta E)$	E_{c} -C ₂	$E_{c}1-S_{4}$	$E_{\rm c}$ 2-S ₄
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	⁴ I _{0/2}	0	11	0	0	${}^{2}G_{7/2}$	18899 (20)	18893	18951	18926
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9/2	104	104	120	156	$(*^{4}G_{2})$	18912 (20)	18926	18977	18979
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		156	156	221	157	(01/2)	18994 (26)	19011	19101	19076
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		228	234	330	257		10004(20) 10030(25)	19044	10124	19070
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		435	438	544	237 561		19030 (23)	19044	19124	19081
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		455	450	544	501	⁴ G		10373	10375	10344
nm pop pop<	4 T	1061	1060	1008	2005	U _{9/2}	-	19323	19373	19344
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1 _{11/2}	1901	1900	1998	2003		19379	19364	19440	19427
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		1997	1992	2064	2040	2	19417	19418	19482	19463
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		2014	2014	2072	2042	K _{13/2}	19479	19476	19524	19554
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		2042	2052	2113	2069		19512	19494	19539	19571
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		2161	2172	2244	2247	4	19521	19522	19583	19573
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		2178	2194	2286	2274	${}^{*}G_{9/2}$	19540	19530	19598	19599
						$^{2}K_{13/2}$	-	19564	19627	19602
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	⁴ I _{13/2}	3924 (8)	3918	3944	3951		-	19600	19672	19621
		3948 (12)	3937	4001	3977		-	19623	19679	19718
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		3970 (10)	3970	4015	3990		19777	19776	19925	19846
		4008 (29)	3993	4070	4017		-	19852	19931	19862
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		4142 (26)	4142	4238	4234					
$ { } { } { } { } { } { } { } { } { } { $		-	4171	4257	4249	$^{2}G_{0/2}$	20923	20936	20983	20967
		4182 (30)	4181	4258	4251	- 9/2	20958	20963	21024	21016
							_	20980	21044	21035
	⁴ I	5853 (9)	5859	5827	5830		_	21000	21052	21038
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	15/2	5903 (13)	5899	5923	5920		21085	21000	21032	21030
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		5045 (10)	5056	5001	5021		21005	21070	2111)	21110
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		3943 (19)	5950	5991	5951	² D	21005	21005	21166	21129
$ {}^{6}F_{1,2} = \begin{array}{ccccccccccccccccccccccccccccccccccc$		-	5988	6094	6027	$D_{3/2}$	21095	21095	21100	21138
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		6234	6227	6314	6322		21146	21118	21181	21141
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		6271 (24)	6259	6355	6373	4				
		-	6305	6453	6419	$G_{11/2}$ and	21239	21242	21306	21269
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		6340	6341	6490	6481	² K _{15/2}	21257	21259	21341	21320
$ \begin{tabular}{l l l l l l l l l l l l l l l l l l l $							21350	21347	21446	21401
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	${}^{4}F_{3/2}$	11415 (16)	11402	11450	11466		21503	21495	21487	21574
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		11477 (31)	11454	11533	11495		-	21519	21545	21588
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$							21543	21550	21597	21633
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	${}^{4}F_{5/2}$ and	12426 (11)	12412	12461	12467		-	21571	21643	21634
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	${}^{2}H_{0/2}$	12454 (15)	12449	12507	12486		-	21595	21671	21664
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	9/2	12509 (16)	12509	12541	12510		_	21606	21704	21700
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		12529 (16)	12546	12585	12590		_	21651	21737	21701
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		12555	12554	12614	12604		_	21679	21769	21725
		12620 (13)	12627	12701	12683		21687	21686	21709	21723
		12684 (14)	12660	12701	12005		21007	21766	21001	21734
$ {}^{4}F_{7/2} = \begin{array}{ccccccccccccccccccccccccccccccccccc$		12004 (14)	12000	12737	12722		21772	21/00	21901	21850
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		12/10	12/14	12621	12790		-	21855	21930	21802
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	4-				10100	2-				
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	${}^{7}F_{7/2}$	13378 (17)	13385	13411	13438	${}^{2}P_{1/2}$	23179 (21)	23186	23252	23241
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	4	13408 (16)	13420	13477	13451	2				
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	${}^{+}S_{3/2}$	13496 (20)	13494	13568	13554	² D _{5/2}	23683	23690	23735	23778
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		-	13501	13579	13557		23739	23749	23807	23802
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	⁴ F _{7/2}	13526	13521	13594	13584		23821	23814	23937	23851
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		13531 (14)	13524	13632	13596					
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$						${}^{2}\mathbf{P}_{3/2}$	-	26117	26161	26183
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	${}^{4}F_{9/2}$	14635 (13)	14650	14713	14712		-	26181	26264	26217
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		14664 (17)	14677	14743	14729					
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		14749 (17)	14748	14818	14808	${}^{4}D_{3/2}$	-	27678	27704	27751
${}^{4}G_{5/2} = \begin{array}{ccccccccccccccccccccccccccccccccccc$		14773 (10)	14781	14822	14827	3/2	27794 (40)	27792	27835	27849
${}^{2} H_{11/2} = 15835 (15) = 15898 = 15949 = 15932 = 27883 (41) = 27919 = 28240 = 279 = 28037 (65) = 28053 = 28050 = 281 = 28057 (52) = 28148 = 28098 = 281 = 28057 (52) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28166 (72) = 28148 = 28098 = 2816 = 28166 (72) = 28166 (72) = 28166 (72) = 28166 (72) = 28148 = 28098 = 281 = 28166 (72) = 28166 (72) = 28164 = 28098 = 2816 = 28166 (72) = 28166 (72) = 28164 = 28008 = 2816 = 28166 (72) = 28164 = 28008 = 2816 = 28166 (72) = 28164 = 28008 = 2816 = 28166 (72) = 28164 = 28008 = 2816 = 28166 (72) = 28164 = 28008 = 28164 = 28008 = 28166 (72) = 28043 = 29037 = 29143 = 29037 = 29143 = 29037 = 29143 = 29037 = 29143 = 2913 = 2914 = 29143 = 29251 = 29252 = 2938 = 29453 = 2944 = -292431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 29492 = 2944 = -29431 = 2944 = -2944 = -2944 = -2944 = -2944 = -2944 = -2944 = -2944 = -2944 = -2944 = -2$		14821 (29)	14832	14919	14870					
${}^{2}\mathrm{H}_{11/2} = \begin{array}{ccccccccccccccccccccccccccccccccccc$		1.021 (2))	11002	1.010	11070	4 D	27883 (41)	27919	28240	27952
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	² H	15835 (15)	15898	15949	15932	25/2	28037 (65)	28053	28050	28150
	н _{11/2} ⁴ G _{5/2}	15872 (15)	15006	15051	15932		28057 (05)	28055	28050	28150
		15806 (0)	15019	15951	15940		28100 (72)	20140	28098	28102
		15890 (9)	15918	15970	13933	40	20462 (60)	20.420	20521	20.475
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		15951 (11)	15933	15997	159/2	$D_{1/2}$	28402 (68)	28430	28531	284/5
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		-	15949	16011	16004	2	200.00	200.50	00051	00000
$ \begin{tabular}{ c c c c c c c c c c c c c c c c c c c$		16021 (22)	15975	16058	16032	-I _{11/2}	28960	28960	28964	29039
							29043	29037	29143	29094
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		16987 (29)	16981	17070	17020		29124	29143	29213	29176
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		17053 (20)	17036	17104	17095		-	29251	29252	29301
		17105 (23)	17104	17166	17144		-	29392	29453	29432
							-	29431	29492	29452
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	${}^{4}G_{7/2}$	17233	17233	17302	17282					
17294 (18) 17294 17383 17352 17411 (36) 17421 17509 17498	$(*^{2}G_{7/2})$	17253	17240	17307	17304					
17411 (36) 17421 17509 17498	. 114/	17294 (18)	17294	17383	17352					
1/711(JU) 1/721 1/JU7 1/720		17411 (36)	17421	17509	17498					



Fig. 2. ${}^{4}F_{3/2}$ non-polarised 10 K photoluminescence of Nd³⁺ in NaBi(WO₄)₂ single crystal. The bars show the contributions of the upper and lower Stark ${}^{4}F_{3/2}$ levels.

parameters, however, if the site symmetry is supposed to be C_2 the number increases to nine real and six complex parameters. In both situations one complex cf parameter, S_4^4 or S_2^2 , respectively, is cancelled by an appropriate choice of the reference axis system.

We have used the very simplified simple overlap model (SOM) [13] to provide starting cf parameters as

$$B_{q}^{k} = \langle r^{k} \rangle \sum_{\mu} \rho_{\mu} \left(\frac{2}{1 \pm \rho_{\mu}} \right)^{k+1} A_{q}^{k}(\mu), \quad \rho_{\mu} = \rho_{0} \left(\frac{R_{0}}{R_{\mu}} \right)^{3.5}$$
(2)

Detailed description of the above expressions can be found elsewhere [13]. The advantage of SOM in comparison to other empirical models lies in the consideration of only the first neighbours of R, with the values of the effective charge for the oxygen and the overlap ρ between orbitals of R and oxygen as the two unique adjustable parameters. The symmetry properties of the Nd³⁺ site and the effective charge attributed to ligands are taken into account through the charge lattice sums, A_q^k , and the radial integrals $\langle r^k \rangle$ for Nd³⁺. ρ varies for each ligand as a function of the distance, R_{μ} , according to the indicated potential law, R_o being the shortest distance. Nd³⁺ is expected to replace Bi³⁺ in the NBW host. In

Nd³⁺ is expected to replace Bi³⁺ in the NBW host. In order to minimise the simulation uncertainty we started with Nd³⁺ fi parameters determined in Na₅Nd(MoO₄)₄ having a similar site symmetry [14], some of them fixed to standard values [11]. Initial cf parameters calculated from SOM assume Nd³⁺ in the crystallographic position of Bi1 or Bi2, both with S_4 symmetry, and the respective neighbours given in Ref. [6]. After convergence of each one of separated fits of the whole observed energy level scheme given in Table 1, the assumption of Nd³⁺ in a Bi2 site provides the lowest root mean square deviation, σ (see Table 2 footnote for definition). The best collection of five real B_q^k and one complex S_q^k cf and fi refined parameters was selected then as the starting set in the calculation performed applying the descent of symmetry method assuming a C₂ site symmetry for Nd³⁺ in NBW. The

Table 2 Free ion and cf parameters (cm $^{-1})$ for Nd $^{3+}$ in NaBi(WO $_{4})_{2}$ crystals a

	C ₂	$1-S_{4}(2b)$	$2-S_{4}(2c)$
E ⁰	23/84 (1)	23553(1)	23537(1)
E^1	23404(1)	4792(7)	23337(1) 4792(7)
E^2	$\frac{47}{2}(7)$	$\frac{4752}{2315}(7)$	$\frac{4752}{2315}(7)$
E^{3}	479 98 (8)	23.13 (2) 479 98 (8)	479.98 (8)
e a	21.90(2)	21.90(2)	21.90 (2)
ß	-686(3)	-686(3)	-686(3)
ρ γ	[1500]	[1500]	[1500]
ĩ	890.4 (6)	890.4 (6)	890.4 (6)
м ^{0b}	2.05 (6)	2.05 (6)	2.05 (6)
P ^{2c}	27.75 (13)	27.75 (13)	27.75 (13)
T^2	[348]	[348]	[348]
T^{3}	30 (2)	30 (2)	30 (2)
T^4	93 (2)	93 (2)	93 (2)
T^6	-283(5)	-283(5)	-283(5)
T^7	326 (6)	326 (6)	326 (6)
T^8	[331]	[331]	[331]
B_0^2	386 (18)	707	278
B_2^2	100 (14)	-	-
B_{0}^{4}	-1011 (33)	-942	-975
B_2^4	229 (42)	-	-
S_{2}^{4}	-90 (59)	_	-
B_{4}^{4}	-664 (33)	951	1100
S_4^4	-795 (26)	-305	79
B_{0}^{6}	-96 (46)	-	-
B_{2}^{6}	-63 (39)	-	-
S_{2}^{6}	-179 (63)	-	-
B_4^6	-717 (23)	1141	1177
S_{4}^{6}	-363 (32)	-254	-243
B_{6}°	-68 (45)	-	-
S_6°	-98 (44)	-	-
\mathbf{S}_2	184	316	125
S_4	605	547	612
S ₆	328	466	472
S _T	411	454	452
I Stark levels	85		
σ	13.6		
Residue	10775.6		

^a Values in parentheses refer to e.s.d. in the indicated parameter. Values in square brackets were not allowed to vary in the parameter fitting. All data are in cm^{-1} units.

 b $M^{0},$ $M^{2},$ M^{4} were constrained by the ratios $M^{2}\!=\!0.5625$ $M^{0},$ $M^{4}\!=\!0.3125$ $M^{0}.$

^c P², P⁴, P⁶ were constrained by the ratios P⁴=0.75 P², P⁶=0.50 P². ^d S_k = {1/(2k+1)[(B^k₀)²+2\Sigma_q[(B^k_q)²+(S^k_q)²]]}^{1/2}S_T = $\left[\frac{1}{3}\Sigma_k S^2_k\right]^{1/2}$.

^e $\sigma = [\Sigma(\Delta_i)^2/(l-p)]^{1/2}$, $\Delta = E_o - E_c$, *l*, number of levels, *p* number of parameters.

simulation yielded an energy level sequence in accordance with observed data, for which a very low $\sigma = 13.6 \text{ cm}^{-1}$ value was obtained. Final results of this adjustment are summarised Table 1, energy levels, and Table 2, fi and cf parameters.

To account for the influence of the two possible sites in S_4 symmetry, we have used the optimised fi parameters of Table 1 together with the cf parameters deduced by the SOM with either the Bi1–O or Bi2–O distances of Ref. [6]. Level energy sequences derived for both sites are also given in Table 1. In both cases, the ground levels have been adjusted to 0 cm⁻¹ through the displacement of E^0 . The energy differences between equivalent levels for both S_4 sites are generally similar to the line-width of the corresponding experimental OA band. It is therefore likely that these two different sites were not resolved in the experimental OA and PL measurements.

Table 2 also includes SOM calculated sets of cf parameters corresponding to both S_4 optical centres. Calculated total crystal field strengths S_T , a way to compare cf effects, especially when the number of involved cf parameters is large, are very similar in all cases, and for the two S_4 centres they have almost the same values. The closeness between S_2 values, which are very sensitive to the magnitude of the electrostatic interactions, i.e. to the Nd– oxygen distances, for C_2 and 2- S_4 simulations, supports the reliability of the used starting set of cf parameters.

However, the simple assumption of two Nd³⁺ centres with S₄ site symmetry and both having parallel principal symmetry axes (expected from the crystallographic model) is inconsistent with the polarisation features observed in OA spectra. For instance, if we assume that the ${}^{4}I_{9/2} \rightarrow {}^{2}P_{1/2}$ transition is only seen in an α spectrum, the ${}^{4}I_{9/2} \rightarrow {}^{4}S_{3/2} + {}^{4}F_{7/2}$ transitions should exhibit three α bands that are not allowed in the π spectrum, this being inconsistent with the experimental result of Fig. 1. This disagreement between experimental results and group theory selection rules is not due to an inaccuracy in the orientation of the crystal. In fact, this problem has been already noticed in the assigning of Raman and IR vibration modes of the NBW host [6].

To explain these results together with the enlargement of experimental line-widths, two possibilities should be considered in the light of the crystal field theory. Either the symmetry of the single Nd³⁺ site in NBW is lower than S_4 , i.e. C_2 [5], with no selection rules expected for OA, or following previous suggestions [6], short-range order and non-random distribution are affecting the Na⁺ and Bi³⁺ environments of Nd³⁺ in NBW. In the latter case, each one of the two optically active Nd^{3+} sites in S_4 symmetry would be inhomogeneously broadened and perhaps with different principal optical axis due to the influence of the Na and Bi short-range environment. To obtain more experimental evidence of the coexistence of the two possible Nd substituted centres, time-resolved spectroscopy and radiative lifetime measurements are currently in progress.

6. Conclusions

Nd doped NaBi(WO₄)₂ crystals can be successfully grown by the Czochralski technique with a Nd concentration in the crystal up to 0.3×10^{20} cm⁻³. The energies of 90 Nd³⁺ Stark levels have been experimentally determined and the influence of the polarisation of the light on the optical absorption has been evidenced. According to the crystallographic models at present available for NBW, the consequences on the OA of the assumption of a single Nd³⁺ centre with C₂ symmetry, or two centres in S₄ site symmetry have been discussed. The latter assumption is consistent with the experimental OA line-widths but further changes in the site symmetry should be assumed to account for the polarisation rules observed. The adjustment of observed crystal field levels has provide a good set of fi and cf parameters for Nd³⁺ in NBW.

Acknowledgements

This work is supported by CICyT under grant numbers 2FD97-0912 and MAT1999-1077. A.M. is supported by CONACYT (Mexico). VV. acknowledges support by a sabbatical grant SAB-1995-0709 of the Spanish Education Ministry.

References

- V.K. Tzunov, V.A. Efremov, Yu.A. Velihkodnyi, Crystallochemistry and Properties of Double Molybdates and Tungstates, Nauka, Leningrad, 1986.
- [2] A.A. Kaminskii, H.J. Eichler, K. Ueda, N.V. Klassen, B.S. Redkin, L.E. Li, J. Findeisen, D. Jaque, J. García-Solé, J. Fernández, R. Balda, Appl. Opt. 38 (1999) 4533.
- [3] K. Nitsch, M. Nikl, C. Barta, D. Schultze, A. Triska, R. Uecker, Phys. Status Solidi (a) 118 (1990) K133.
- [4] A.A. Kaminskii, S.N. Bagayev, K. Ueda, H. Nishioka, Y. Kubota, X. Chen, A. Kholov, Jpn. J. Appl. Phys. 34 (1995) L1461.
- [5] J. Hanuza, M. Maczka, J.H. van der Maas, J. Solid State Chem. 117 (1995) 177.
- [6] J. Hanuza, A. Benzar, A. Haznar, M. Maczka, A. Pietraszko, J. H van der Maas, Vib. Spec. 12 (1996) 25.
- [7] V. Volkov, C. Zaldo, J. Cryst. Growth 206 (1999) 60.
- [8] P.V. Klevtsov, V.A. Vinokurov, R.F. Klevtsova, Kristallografiya 18 (1973) 1192.
- [9] V. Volkov, M. Rico, A. Méndez-Blas, C. Zaldo, J. Phys. Chem. Solids (in press).
- [10] D. Garcia, M. Faucher, J. Chem. Phys. 90 (1989) 5280.
- [11] W.T. Carnall, G.L. Goodman, K. Rajnak, R.S. Rana, J. Chem. Phys. 90 (1989) 3443.
- [12] B.G. Wybourne, Spectroscopic Properties of Rare Earth, Wiley, New York, 1965.
- [13] P. Porcher, M. Couto dos Santos, O. Malta, Phys. Chem. Chem. Phys. 1 (1999) 397.
- [14] E. Antic Fidancev, C. Cascales, M. Lemaitre Blaise, P. Porcher, J. Alloys Comp. 207/208 (1994) 178.